

(S)-2-(2-Chloroquinolin-3-yl)-2-[(S)- α -methylbenzylamino]acetonitrile

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Key indicators

Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$
R factor = 0.055
wR factor = 0.093
Data-to-parameter ratio = 14.4

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

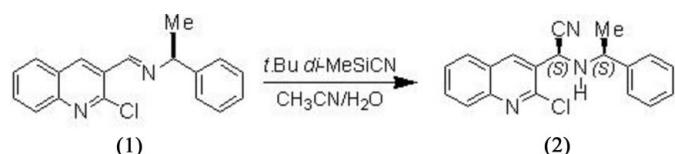
The title compound, $C_{19}H_{16}ClN_3$, crystallizes with two independent molecules in the asymmetric unit. The structure is stabilized by $\text{C}-\text{H}\cdots\text{N}$, $\text{N}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{Cl}$ hydrogen bonds.

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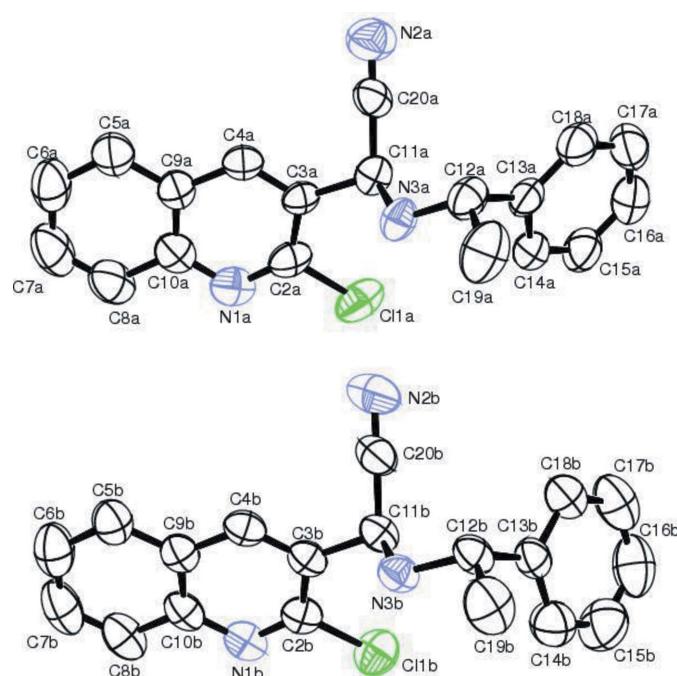
Comment

Quinolines are an important group of heterocyclic compounds. Among these, 2-chloro-3-formylquinolines occupy a prominent position as they are key intermediates for further (β)-annelation of a wide variety of rings and for various functional group interconversions (Meth-Cohn, 1993). Particular interest in quinoline derivatives arises owing to their biological activity, namely as antibiotics (Jackson & Meth-Cohn, 1995; Kansagra *et al.*, 2000), anti-inflammatories (Schroderet, 1989), anti-tumourals (Joseph *et al.*, 2002), anti-oxidants (Laalaoui *et al.*, 2003) and analgesics (Heide *et al.*, 1986; Solomon, 1970). In the same way, α -aminoacids are of great biological and economic importance (Williams, 1989). The asymmetric Strecker reaction is one of the most important methods for the synthesis of enantiomerically pure α -amino-nitrile derivatives, which are useful intermediates for the synthesis of α -aminoacids. The use of (S)-(-)- α -methylbenzylamine-derived aldimines has a significant role in the diastereoselective Strecker synthesis (Bhanu-Prasad *et al.*, 2004).

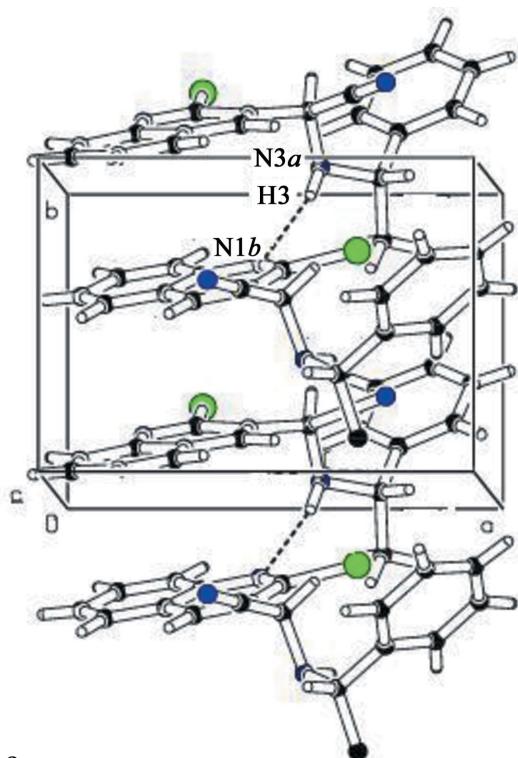
In recent years, we have developed a programme devoted to the synthesis and biological evaluation of quinolyl derivatives (Moussaoui *et al.*, 2002; Kedjadja *et al.*, 2004; Menasra *et al.*, 2004, 2005; Rezig *et al.*, 2000). In a continuation of our efforts in this area, we report here a short and efficient procedure for the preparation of the title α -aminonitrile, (2), containing a quinolyl ring system, and its crystal structure determination.



The crystallographic asymmetric unit of (2) contains two independent molecules, labelled *a* and *b* (Fig. 1). The analysis shows that atoms C11 and C12 each have an *S* configuration in both independent molecules. The geometric parameters of (2) (Table 1) are in agreement with those of other structures containing similar molecular connectivity (Benali-Cherif, Cherouana *et al.*, 2002; Benali-Cherif, Dokhane & Abdaoui, 2002). The 11 atoms defining the chloroquinolyl planes, *i.e.* N1 and Cl1–C10, have maximum deviations of 0.0037 (3) Å for

**Figure 1**

Views of the two independent molecules of (2), showing 50% probability displacement ellipsoids and the atom-numbering scheme. H atoms have been omitted for clarity.

**Figure 2**

The unit-cell contents of (2), highlighting the $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonding (dashed lines).

atom $\text{C}7a$ and 0.0035 (3) Å for atom $\text{C}7b$ for the two independent molecules. In the crystal structure, these planes are almost parallel, forming a dihedral angle of 7.40 (4)°. The clear difference between molecules *a* and *b* is noted in the dihedral

angles formed between the chloroquinolyl and phenyl groups of 27.83 (6) and 50.53 (8)°, respectively.

The three-dimensional crystal structure of (2) is stabilized via a variety of hydrogen-bonding interactions of the types $\text{C}-\text{H}\cdots\text{N}$, $\text{N}-\text{H}\cdots\text{N}$ and $\text{C}-\text{H}\cdots\text{Cl}$ (Steiner, 1996), as analysed by PARST (Nardelli, 1995) and summarized in Fig. 2 and Table 2.

Experimental

Chiral imine (1) was prepared by condensation of optically active (*S*)(*-*)- α -methylbenzylamine with 2-chloro-3-formylquinoline, according to the literature procedure of Meth-Cohn *et al.* (1981). Treatment of (1) with *tert*-butyldimethylsilyl cyanide at room temperature in CH_3CN solution with a few drops of water provided a mixture of two diastereoisomers as a yellow solid (yield 84%). Crystals of (2) were obtained by fractional crystallization from a hexane– CH_2Cl_2 (9:1) solution of this mixture. The isomeric ratio of (2) (63%) was determined from the ^1H NMR spectrum of the crude product (m.p. 383 K).

Crystal data

$\text{C}_{19}\text{H}_{16}\text{ClN}_3$
 $M_r = 321.8$
Monoclinic, $P2_1$
 $a = 9.8540$ (1) Å
 $b = 7.1090$ (1) Å
 $c = 23.9330$ (3) Å
 $\beta = 91.590$ (2)°
 $V = 1675.91$ (4) Å³
 $Z = 4$

$D_x = 1.275$ Mg m⁻³
Mo $K\alpha$ radiation
Cell parameters from 2548 reflections
 $\theta = 1.7\text{--}28.0^\circ$
 $\mu = 0.23$ mm⁻¹
 $T = 293$ (2) K
Prism, yellow
 $0.2 \times 0.15 \times 0.1$ mm

Data collection

Nonius KappaCCD area-detector diffractometer
 φ and ω scans
Absorption correction: none
35588 measured reflections
7662 independent reflections

4482 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.065$
 $\theta_{\text{max}} = 28.0^\circ$
 $h = -13 \rightarrow 12$
 $k = -8 \rightarrow 9$
 $l = -31 \rightarrow 31$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.093$
 $S = 1.03$
7662 reflections
531 parameters
H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0312P)^2 + 0.1354P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 0.15$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.15$ e Å⁻³
Absolute structure: Flack (1983),
3303 Friedel pairs
Flack parameter: -0.01 (5)

Table 1
Selected geometric parameters (Å, °).

N2a—C20a	1.113 (3)	N2b—C20b	1.134 (3)
C17a—C18a	1.374 (4)	C17b—C18b	1.394 (5)
N3a—C11a—C20a	114.4 (2)	N3b—C11b—C20b	109.6 (2)
C11a—N3a—C12a	117.1 (2)	C11b—N3b—C12b	114.8 (2)
C3a—C11a—N3a—C12a	-169.6 (2)	C3b—C11b—N3b—C12b	178.9 (2)
C4a—C3a—C11a—N3a	100.8 (3)	C4b—C3b—C11b—N3b	89.2 (3)
C19a—C12a—C13a—C18a	-99.7 (3)	C19b—C12b—C13b—C18b	-125.7 (3)

Table 2
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N}3a-\text{H}3\cdots \text{N}1b^i$	0.90 (3)	2.31 (3)	3.189 (3)	165.6 (2)
$\text{N}3b-\text{H}3'\cdots \text{N}2a$	0.87 (2)	2.68 (2)	3.508 (3)	159.6 (19)
$\text{C}11b-\text{H}11'\cdots \text{Cl}1b$	0.98 (2)	2.641 (19)	3.067 (3)	106.3 (13)
$\text{C}11a-\text{H}11\cdots \text{Cl}1a$	1.00 (3)	2.76 (2)	3.044 (3)	97.4 (14)

Symmetry code: (i) $x, y - 1, z$.

All H atoms were freely refined except for the methyl H atoms bonded to atom C19, for which $\text{C}-\text{H} = 0.96 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{iso}}(\text{C})$.

Data collection: *KappaCCD Server Software* (Nonius, 1998); cell refinement: *DENZO* and *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* and *SCALEPACK*; program(s) used to solve structure: *SIR2004* (Burla *et al.*, 2005); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *enCIFer* (Allen *et al.*, 2004).

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